Functional Materials

MS.3.P062 In situ observation of the electron beam induced phase transformation of CaCO₃ to CaO as a function of the accelerating voltage (20-300kV)

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CaCO₃ (calcite) is one of the important systems in the field of biomineralisation and the most common carbonate of the earth crust inevitable for the manufacture of cements. Limestone is the chief raw material, which when heated about 900° C forms CaO (lime) by the reaction: CaCO₃ \rightarrow CaO+CO₂ \uparrow . Apparently CaCO₃ undergoes the same phase transformation to CaO by electron irradiation.

Therefore the new challenge arises to search for conditions and instrumental settings especially the applied accelerating voltage to study in situ the phase transformation on an atomic scale.

The accelerating voltage was reduced from 300 kV down to 20 kV by using a CM20 operating at 200 kV, equipped with a LaB6 cathode, a TITAN operating at 300 kV and 80 kV equipped with a field emission gun and an imaging-side Cs-corrector and the newly developed SALVE prototype microscope equipped with a field emission gun, an image-side Cs-corrector and a monochromator [1] operating at 20, 40 and 80 kV. At high accelerating voltages the radiation damage of CaCO₃ is starting immediately with the electron-radiation and can be divided up in three stages first the amorphisation of the crystalline structure, then producing holes and at least recrystalisation in a polycrystalline structure with significant volume and correlated mass loss during the reaction time. Figure 1 a), b) show a CaCO₃ crystal and the correlated FFT, where the lattice planes can be imaged without any radiation damage. Figure 1 c), d) show the same crystal after 25 min radiation with a dose rate of $4x10^4$ e/nm²s, where a polycrystalline structure is grown with lattice spacing fitting the CaO structure. Exemplary a starting image taken with the Titan and a dose rate of $4x10^4$ e/nm²s at 300kV accelerating voltage is presented in Figure 1 e), where the radiation damage results in fast hole ptoduction. However, the crystal structure of the original calcite is still present in the FFT of the cut-out. We found that reduced accelerating voltages below 80 kV slows down the radiation damage and gives the freedom to align the instrument to obtain information on the atomic scale before significant beam damage effects appear. The quantitative studies of volume-loss at 20, 40 and 80 kV delivered increasing volume loss with decreasing accelerating voltages. This effect at low accelerating voltages is counterbalanced by increasing contrast which allows the use of decreasing dose rates. To determine the mass loss quantitatively, the chemistry has to be known, therefore additionally EELS spectra were taken. Figure 2 show spectra taken with the SALVE microscope at 80kV a) of the C K, Ca L₂₃ and b) O K-edges where the fading out signal of carbon confirms the phase transformation. The high stability of the instrument enables additionally to measure in situ the change of Ca ELNES during the phase transformation, which is correlated to bond length variation of Ca. The undistorted octahedral coordination of calcium with oxygen as binding partner is preserved, where solely the bond length change from 2,357 Å to 2,407 Å by 5 pm. As a result of the octahedral coordination the degenerated energy levels of the 3d shell split up into the t_{2g} and e_g energy levels by Δ . The value of Δ is influenced by the distances R between the cation in our case Ca and the surrounding ligands and can be described by: $\Delta = Q/R^5$ where Q is approximately a constant for cations of similar valence in the same transition series and identical charge on the ligands [2]. Finally we obtained for the ratio of energy level splitting for CaCO₃ and CaO Δ_{CaO} / Δ_{CaCO3} experimentally 0,82 which agrees nicely with the theoretically prediction of 0,9.

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Figure 1. In situ study of calcite recorded with the SALVE at 20 kV (a-d) and Titan at 300kV (e,f) accelerating voltage and the corresponding FFTs, which shows the change of a single crystal structure to a polycrystalline structure after electron-irradiation. Please describe what we see in a,- f in more detail.



Figure 2. Electron energy loss spectra taken at 80kV during one phase transformation series of the a) C K, Ca L_{23} and b) O K-edges.